WHAT IS CLAIMED IS:

1	1.	A method of making a lithiated manganese dioxide for a primary lithium
2	battery comprising:	
3		contacting a manganese dioxide with a lithium ion source at a lithiation
4	temperature s	sufficient to substantially replace protons in the manganese dioxide with lithium
5	ions; and	
6		heating the manganese dioxide at a water removal temperature sufficient to
7	substantially	remove residual and surface water to produce a lithiated manganese dioxide
8	having an X-	ray diffraction pattern substantially similar to the X-ray diffraction pattern of the
9	manganese d	ioxide prior to lithiation.
1	2.	The method of claim 1, wherein the manganese dioxide is persulfate derived
2	chemical manganese dioxide.	
1	3.	The method of claim 1, wherein the manganese dioxide is gamma-manganese
2	dioxide.	
1	4.	The method of claim 1, wherein the lithium ion source is an aqueous solution
2	including a li	thium salt.
1	5.	The method of claim 4, wherein the lithium salt is a lithium hydroxide.
1	6.	The method of claim 1, wherein the lithiation temperature is between 40 C
2	and 100 C.	
1	7.	The method of claim 1, wherein the water removal temperature is between
2	180 C and 500 C.	
1	8.	The method of claim 1, wherein the water removal temperature is between
2	200 C and 46	0 C.
1	9.	A method of making a cathode for a battery comprising:
2		contacting a manganese dioxide with a lithium ion source;
3		heating the manganese dioxide to produce a lithiated manganese dioxide

having an X-ray diffraction pattern substantially similar to the X-ray diffraction pattern of the 4 manganese dioxide prior to lithiation; and 5 6 coating a current collector with a composition including a carbon source, and the cathode active material, wherein the cathode active material includes a manganese 7 dioxide. 8 10. 1 The method of claim 9, wherein the manganese dioxide is persulfate derived chemical manganese dioxide. 2 1 11. The method of claim 9, wherein the manganese dioxide is gamma-manganese dioxide. 2 12. 1 The method of claim 9, wherein the lithium ion source is an aqueous solution 2 including a lithium salt. 13. The method of claim 12, wherein the lithium salt is a lithium hydroxide. 1 14. The method of claim 9, wherein the lithiation temperature is between 40 C 1 and 100 C. 2 15. The method of claim 9, wherein the water removal temperature is between 1 180 C and 500 C. 2 1 16. The method of claim 9, wherein the water removal temperature is between 200 C and 460 C. 2 17. 1 A primary lithium battery comprising: 2 an anode including a lithium-containing anode active material; 3 a cathode including a lithiated manganese dioxide having an X-ray diffraction 4 pattern substantially similar to the X-ray diffraction pattern of the manganese dioxide prior to 5 lithiation; and a separator between the anode and the cathode. 6

The battery of claim 17, wherein the lithium-containing anode active material

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is lithium or a lithium alloy.

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- 1 19. The battery of claim 17, further comprising a nonaqueous electrolyte in contact with the anode, the cathode and the separator.
- 1 20. The battery of claim 19, wherein the nonaqueous electrolyte includes an organic solvent.
- The battery of claim 17, wherein the battery has high current capability and discharge capacity greater than a lithium-manganese dioxide battery including heat treated manganese dioxide.